

An atomistic investigation on the wear of diamond during atomic force microscope tip-based nanomachining of Gallium Arsenide

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Abstract

This paper investigated the wear mechanism of diamond during the atomic force

microscope (AFM) tip-based nanomachining of Gallium Arsenide (GaAs) using molecular dynamics (MD) simulations. The elastic-plastic deformation at the apex of the diamond tip was observed during the simulations. Meanwhile, a transition of the diamond tip from its initial cubic diamond lattice structure sp^3 hybridization to graphite lattice structure sp^2 hybridization was revealed. Graphitization was, therefore, found to be the dominant wear mechanism of the diamond tip during the nanometric cutting of single crystal gallium arsenide for the first time. The various stress states, such as hydrostatic stress, shear stress, and von Mises stress within the diamond tip and the temperature distribution of the diamond tip were also estimated to find out the underlying mechanism of graphitization. The results showed that the cutting heat during nanomachining of GaAs would mainly lead to the graphitization of the diamond tip instead of the high shear stress-induced transformation of the diamond to graphite. The paper also proposed a new approach to quantify the graphitization conversion rate of the diamond tip.

Keywords: AFM tip-based nanomachining; MD simulation; Diamond tip wear; Single crystal gallium arsenide; Graphitization

1. Introduction

Gallium arsenide (GaAs), a high value-added semiconductor material, has unique advantages in contrast to silicon, such as higher electron mobility, wider bandgap and higher temperature resistance [1]–[3]. It has been broadly used in the field of microelectronics and optoelectronics to make smart devices for telecommunication,

fibre-optic communication, satellite communication, cable television, vehicle navigation system and semiconductor light-emitting diodes [4]–[7]. The elastic modulus and fracture toughness of GaAs is much lower than single crystal silicon [8], which makes it much harder to machine than silicon [9].

AFM tip-based nanomachining of GaAs offers a low-cost solution to perform flexible direct writing [10] with nanometer scale accuracy in a single pass under various conditions, such as ultra-vacuum, liquid and low temperature or high temperature [11]. For example, Hyon et al. [12] employed AFM tip-based nanomachining approach to fabricate patterns as narrow as 10 nm on the GaAs workpiece. Schumacher et al. [13] obtained a single-electron transistor by fabricating a GaAs/AlGaAs heterostructure surface. Additionally, AFM tip-based nanomachining approach has been successfully applied to generate nanodots, nanolines, two-dimensional (2D), and even three-dimensional (3D) nanostructures with serval nanoNewton force [14]–[18]. This is particularly useful for the fabrication of 3D GaAs nanostructures, for instance, the hemisphere cavity for light-emitting diodes quantum devices on the GaAs substrate. Due to the nature of mechanical material removal, AFM tip-based machining outperforms the focussed ion beam (FIB) machining approach [19] in terms of high material removal rate and consistent device performance due to no ion contamination on the fabricated devices.

The diamond succumbs to wear during nanomachining process, leading to the worsening of the machined surface by the virtue of the transition of machining mode from ductile-regime to brittle fracture [20]. Catastrophic wear of diamond tip leads to

stress concentration in the cutting zone which brings about microcracks and fractures. Correspondingly, an understanding of the wear mechanism of the diamond tip is necessary to mitigate the wear of diamond tip during high-volume production of GaAs-enabled devices. This is the motivation of the research paper. Single crystal diamond and polycrystalline diamond tools have been experimentally observed to undergo different wear mechanisms, such as diffusion wear [21], thermo-chemical wear [22], abrasive wear [23], graphitization [24], and chemical diffusion [25]. Nevertheless, experimental technologies are currently limited by the timescales to monitor real-time events such as the wear of material occurring within a few picoseconds where simulations can contribute very strongly as accelerated digital nanomanufacturing tools. Correspondingly, molecular dynamics (MD) simulation has become a favourable choice which is capable of revealing many additional atomistic insights of the tribological phenomena in terms of energy and structural changes [26]. For example, carbon atoms of the diamond tool were observed to transform to graphite structure during single point diamond turning of mold steel by using MD simulation [27]. Goel et al. employed MD simulation to reveal sp^3 - sp^2 hybridization order-disorder transformation of carbon atoms when cutting silicon [28] and silicon carbide [29]. Additionally, the diffusion wear of the diamond tool was shown in the study of machining aluminum alloy [30] by utilizing MD simulations. These works have proved that MD simulation is an effective technique to unravel the wear mechanism of diamond cutting tools. However, the wear mechanism of the diamond tool during nanometric cutting of single crystal GaAs has not been revealed yet. Therefore, this paper

investigates the wear mechanism of the diamond tip using large scale MD simulations. The innovation of this paper is to simulate the atomic force microscope tip-based nanomachining process on high brittleness single crystal GaAs material to investigate the wear mode of the diamond tip. It is different from the previous simulation works on hard-brittle materials, such as silicon carbide and silicon, in terms of the geometry of the tool top mimicking an AFM tip and the use of the type of potential function. The potential influential factors on the wear of diamond tips, such as thermal effect and high shear stress, were analyzed and discussed in the subsequent sections. Additionally, a quantitative assessment of the graphitization rate of the diamond tip was also proposed.

2. MD simulation

2.1. MD simulation model

In this study, a 3D MD simulation model for AFM tip-based nanomachining of single crystal GaAs workpiece was established. The simulation was implemented in the open-source code, "Large-scale atomic/molecular massively parallel simulator" LAMMPS [31]. As shown in Fig. 1, a square pyramid diamond tip with the (1 0 0) crystallographic orientation was considered as a non-rigid body with a 1:1 aspect ratio. The tip was used to cut the GaAs on the (010) orientation in the $\langle 100 \rangle$ direction. Since the focus of the investigation here is to understand the wear of diamond, the anisotropy aspects of the tool and the workpiece were not reported here. The GaAs workpiece was built with a

size of 31.0 nm × 10.1 nm × 13.4 nm in the X (length), Y (width), and Z (height) directions, respectively. The green and blue colour atoms represent the boundary layers and thermostat layers of the tool and workpiece while red and purple colour belongs to gallium and arsenide atoms, respectively, within the GaAs workpiece Newton layers and the yellow atoms represent the Newton atoms of the diamond tool. The thermostat layer was prescribed a Berendsen thermostatic dynamics to dissipate the artificial Joule heat generated during the cutting process.

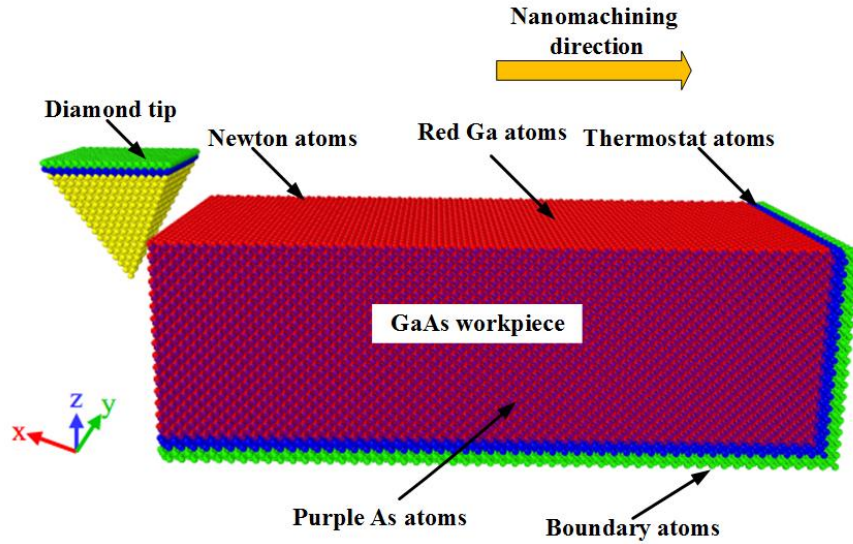


Fig. 1. The MD simulation model for AFM diamond tip-based nanomachining of single crystal GaAs

During the simulation, the initial temperature of the GaAs workpiece and diamond tip machining tool were equilibrated at 300 K by using the Nose-Hoover method [32]. Also, atoms in the X, Y, and Z direction were set into shrink-wrapped, shrink-wrapped, and periodic boundary conditions, respectively. The shrink-wrapped boundary condition is capable of wrapping the atoms dropped from the diamond tip while there will be no new atoms at the corresponding position to supplement, which is very suitable to

investigate the change or transformation of tip structure during nanomachining. Finally, the simulation of cutting experiments was carried out by prescribing a depth of cut of 3 nm and length of cut of 20 nm (tip velocity of 200 m/s) under a fixed number of atoms (N), volume (V), and energy (E), so-called NVE microcanonical ensemble. The simulation parameters are shown in detail in Table 1. The simulation results were visualised and analyzed using visual molecular dynamics (VMD) [33] and Open Visualization Tool (OVITO) software [34].

Table 1

Simulation parameters of MD simulation model

Substrate dimensions	31.0 nm × 10.1 nm × 13.4 nm
Number of GaAs substrate atoms	190080
Machining tool	Diamond tip (wearable)
Number of diamond tool atoms	13107
Crystallographic plane of diamond tip	(1 0 0)
GaAs lattice constant	5.65 Å
Diamond lattice constant	3.57 Å
Diamond tip aspect ratio	1:1
Depth of cut	3 nm
Machining distance	20 nm
Cutting velocity	200 m/s
Crystallographic plane of GaAs substrate	(0 1 0)
Equilibration temperature	300 K
Timestep	1 fs

2.2. Selection of potential functions

In this work, the force field consisted of a hybrid scheme such that the Tersoff potential function (C-C interactions), updated Bond-Order Potential (BOP) function (Ga-As interactions) and Ziegler-Biersack-Littmark (ZBL) potential function (for cross interaction between carbon with GaAs atoms) were respectively employed. The classical Tersoff potential function [35] was selected to describe the C-C interactions in the diamond tip machining tool [36]. The calculations of a Tersoff potential function are shown in Eq. (1-8) [37].

$$E = \sum_{i=j} V_{ij} \quad (1)$$

$$V_{ij} = f_c(r_{ij})[f_R(r_{ij}) + b_{ij}F_A(r_{ij})] \quad (2)$$

$$f_R(r_{ij}) = A \exp(-\lambda r_{ij}) \quad (3)$$

$$f_A(r_{ij}) = -B \exp(-\mu r_{ij}) \quad (4)$$

$$f_c(r_{ij}) = \begin{cases} 1 & r_{ij} < R \\ \frac{1}{2} + \frac{1}{2} \cos \left[\pi \frac{r_{ij}-R}{S-R} \right] & R < r_{ij} < S \\ 0 & r_{ij} > R \end{cases} \quad (5)$$

$$b_{ij} = (1 + \beta_i^n \delta_{ij}^n)^{-1/2n} \quad (6)$$

$$\delta_{ij} = \sum_{k \neq i,j} f_c(r_{ik}) \mathcal{G}(\theta_{ijk}) \quad (7)$$

$$\mathcal{G}(\theta_{ijk}) = \frac{(1+c^2)/(d^2-c^2)}{[d^2+(h-\cos \theta_{ijk})^2]} \quad (8)$$

where E represents the system total energy, the V_{ij} refers to bond energy in the whole atomic bonds, i, j , and k mean the atomic label of the system, f_c is a smooth cutoff function to restrain the range of the potential, f_R is a two-body term, r_{ij} represents the length function of the i and j bond, b_{ij} means the bond order term, F_A consists of the

three-body interactions, f_A and f_c are the attractive pair potential and smooth cutoff function, respectively. δ_{ij} refers to the atom number in which the atoms i besides the ij bond. The θ stands for the bond angle.

A BOP function based on quantum mechanical theory [38][39] is capable of predicting the structural changes and binding energy trends in the GaAs system, it was adopted to describe the interaction of Ga-Ga, As-As, and Ga-As, which was computed by Eq. (9). The GaAs bond-dependent BOP parameters are shown in Table 2 [40]. The repulsive and bond integral terms were calculated by Goodwin-Skinner-Pettifor (GSP) function [41].

$$E = \frac{1}{2} \sum_{i=1}^N \sum_{j=i_1}^{i_N} \varphi_{ij}(r_{ij}) - \sum_{i=1}^N \sum_{j=i_1}^{i_N} \beta_{\sigma,ij}(r_{ij}) \cdot \theta_{\sigma,ij} - \sum_{i=1}^N \sum_{j=i_1}^{i_N} \beta_{\pi,ij}(r_{ij}) \cdot \theta_{\pi,ij} + U_{prom} \quad (9)$$

where the r_{ij} is the distance function for atoms i and j , $\varphi_{ij}(r_{ij})$ is the repulsive energy and short-range two-body function, $\beta_{\sigma,ij}(r_{ij})$, and $\beta_{\pi,ij}(r_{ij})$ refer to bond integrals, $\theta_{\sigma,ij}$ and $\theta_{\pi,ij}$ are specific bond-orders, U_{prom} means the promotion energy about sp-valent systems. Following the BOP potential function, the cutoff was modified to 11.1 Å which could effectively speed up the inter-processor communication.

Table 2

The GaAs bond-dependent BOP parameters [40]

Symbol	Quantity	Ga-Ga	As-As	Ga-As
r_0	GSP reference radius (Å)	2.4235	2.1200	2.3800
r_c	GSP characteristic radius (Å)	2.4235	2.1200	2.3800
m	GSP attractive exponent	1.4509	1.3059	1.9652
n	GSP repulsive exponent	0.7255	0.6529	0.9826
n_c	GSP decay exponent	2.6234	2.6304	2.6234

r_l	Spline start radius (Å)	3.0000	3.0000	3.0000
r_{cut}	Spline cutoff radius (Å)	3.7000	3.6500	3.7000
φ_0	Repulsive energy prefactor (eV)	1.5520	3.9800	2.1000
$\beta_{\sigma,0}$	σ bond integral prefactor (eV)	1.5233	3.0877	1.7959
$\beta_{\pi,0}$	π bond integral prefactor (eV)	0.0975	0.9263	0.3233
c_σ	Empirical θ_σ parameter	1.5193	3.6016	0.8534
c_π	Empirical θ_π parameter	1.0000	1.0000	1.0000
f_σ	Band filling fraction ($e^-/8$)	0.4456	0.6558	0.5000
k_σ	Skewing prefactor	-25.6485	0.7600	0.0000

Additionally, the ZBL potential function [42] was employed to describe the interaction of carbon with the GaAs substrate as shown in Eq. (10-11).

$$E_{ij}^{ZBL} = \frac{1}{4\pi\epsilon_0} \frac{Z_i Z_j e^2}{r_{ij}} \varphi(r_{ij}/a) + S(r_{ij}) \quad (10)$$

$$a = \frac{0.46850}{Z_i^{0.23} + Z_j^{0.23}} \quad (11)$$

where ϵ_0 refers to the electrical permittivity of vacuum, e means the electron charge, Z_i and Z_j are the nuclear charges of the two types of atom, r_{ij} is identical to mentioned above, $S(r_{ij})$ is a switching function which can limit the energy, force, and curvature down to zero smoothly within cutoff range. Moreover, the inner cutoff and outer cutoff were set to 3.0 Å and 4.0 Å, respectively.

2.3. Parametric analysis

In this study, Eq. (12) [43]–[45] was employed to compute the temperature of the diamond tip by conversion of average kinetic energy:

$$K.E. = \frac{3}{2} N K_b T \quad (12)$$

where $K.E.$ refers to the kinetic energy of the GaAs substrate atoms, N is the number of atoms, K_b is the Boltzmann constant.

In addition, the virial stress computation method [36] was used to compute the stress components of the GaAs substrate and diamond tip. It can be described as follows [46]–[48]:

$$\sigma_{ab}(i) = \frac{1}{N_A} \sum_i \left[\frac{m_i v_i^a v_i^b}{V_i} + \frac{1}{2V_i} \sum_j F_{ij} \frac{x_{ij}^a x_{ij}^b}{r_{ij}} \right] \quad (13)$$

where σ_{ab} refers to six direction symmetric stress components of each atom, ab represents x, y, and z, N_A is the number of atoms in an area A , m_i is the mass of the i atom, v_i^a and v_i^b mean the a component and b component of the i atom velocity, V_i is the volume of the i atom, F_{ij} is the force between the i atom and the j atom, x_{ij}^a and x_{ij}^b represent a direction vector and b direction vector of i atom and j atom, r_{ij} is the distance between the i atom and the j atom.

The hydrostatic stress (average of three principal stresses) and von Mises stress of GaAs substrate and diamond tip were further investigated, which could be calculated by Eq. (14) and Eq. (15), respectively [49]–[51]. In this case, the hydrostatic stress and von Mises stress were averaged over a 20Å box for each GaAs substrate atom.

$$\sigma_{hydro}(i) = \frac{1}{3} [\sigma_1 + \sigma_2 + \sigma_3] \quad (14)$$

$$\sigma_{von}(i) = \left\{ \frac{1}{2} \left[\left(\sigma_{xx}(i) - \sigma_{yy}(i) \right)^2 + \left(\sigma_{yy}(i) - \sigma_{zz}(i) \right)^2 + \left(\sigma_{zz}(i) - \sigma_{xx}(i) \right)^2 + 6 \left(\sigma_{xy}^2(i) + \sigma_{zy}^2(i) + \sigma_{xz}^2(i) \right) \right] \right\}^{1/2} \quad (15)$$

3. Results and discussion

3.1 Deformation and wear of diamond tip

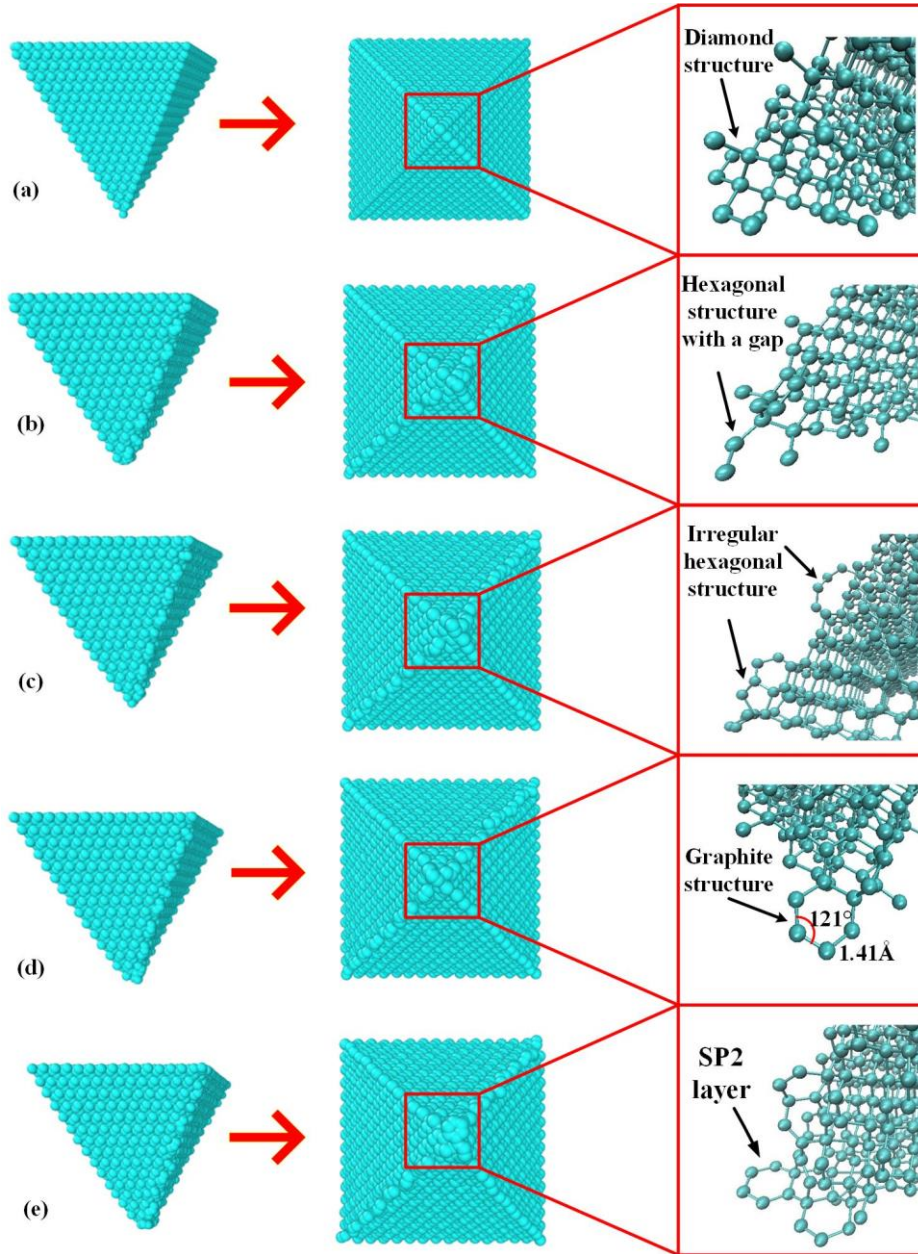


Fig. 2. The observation of structural change of the diamond tip during nanometric cutting of GaAs. (a) initial diamond tip structure. (b-e) the diamond tip structures at the cutting distances of 2 nm, 6 nm, 10 nm, and 20 nm, respectively.

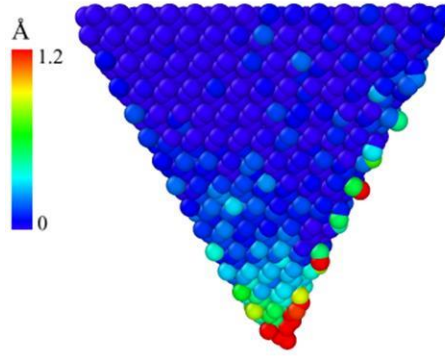


Fig. 3. The atomic displacement of the carbon atoms in the diamond tip after a nanomachining distance of 10 nm

As shown in Fig. 2, the diamond tip experienced elastic-plastic deformation during the nanomachining process caused by a sp^3 - sp^2 hybridization transition. The deformation hotspot appeared at the apex of the diamond tip, meanwhile, the cutting edge of the diamond tip showed incipient deformation. Furthermore, from Fig. 3, it can be observed that the maximum atomic displacement of the apex of the diamond tip could reach 0.12 nm. The occurrence of diamond tip elastic-plastic deformation behavior is attributed to high pressure acting on the diamond tip structure [52][53].

Moreover, from the enlarged diagrams of the apex of the diamond tip shown in Fig. 2, it is apparent that a hexagonal structure with two C-C bond gaps (as shown in Fig. 2(b)) forms at the cutting distance of 2 nm. With the nanomachining advancement, an irregular hexagonal structure, as illustrated in Fig. 2(c), was observed at the cutting distance of 6 nm. Then, a complete hexagonal carbon-cycle structure occurred at the distance of 10 nm, as demonstrated in Fig. 2(d). Subsequently, the measured bond angle and bond length of C-C were 121° and 1.41 Å, respectively. This is very close to the graphite hexagonal lattice structure with 121° C-C bond angle and 1.42 Å C-C bond length [29][54]. Finally, at the cutting distance of 20 nm, the presence of a clear sp^2

graphite layer structure was observed, as seen in Fig. 2(e). This phenomenon suggests that the apex of the diamond tip transformed from an initial sp^3 hybridization diamond lattice structure to sp^2 graphite lattice structure due to the C-C bond transformation and reorganization of bonds. From this, we can infer that the graphitization governs the wear mechanism of diamond tip during AFM tip-based nanomachining of GaAs. Correspondingly, thermal and shear stresses were analyzed in the subsequent sections to investigate the underlying mechanism of graphitization of the diamond tip thoroughly. Further advancement of the machine leads to the graphite layer of diamond to fall off and separated from the diamond tip and create a small concave structure, previously reported by Cheng *et al.* [55].

3.2 Structural transformations in GaAs workpiece

Various stress states in the workpiece were calculated to see if they can induce classical thermodynamic phase transition [47]. As illustrated in Fig. 4, the average hydrostatic stress in the cutting zone experienced a rapid upward trend reaching the maximum scale at ~1.3 GPa and kept a relatively fluctuation between 0.4 GPa and 0.6 GPa during the nanomachining process. Furthermore, the peak hydrostatic stress of about 17.5 GPa occurred in the primary shear slip zone, as shown in Fig. 5.

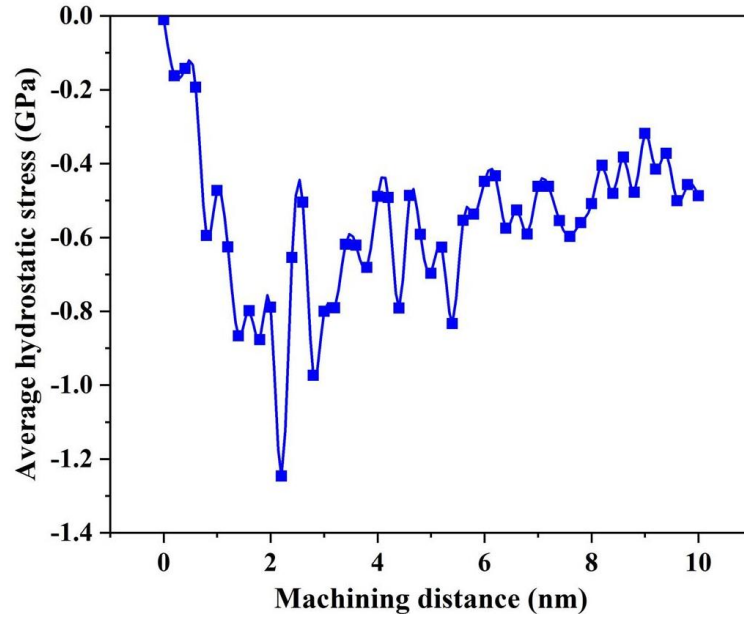


Fig. 4. Variation in the average hydrostatic stress in the cutting zone

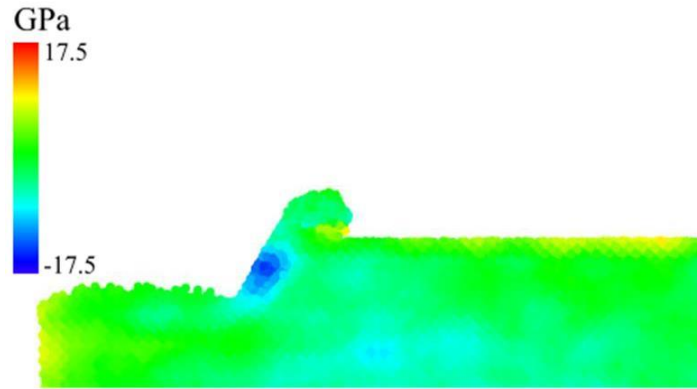


Fig. 5. the distribution of the hydrostatic within the machined single crystal GaAs workpiece.

The peak stress was consistent with the recent experimental reported limits indicating that the single crystal GaAs could well transform from a zinc blende structure (GaAs-I) to rocksalt structure (GaAs-II) [56]. To affirm this the coordination number (CN) was computed during the MD simulations, as indicated in Fig. 6.

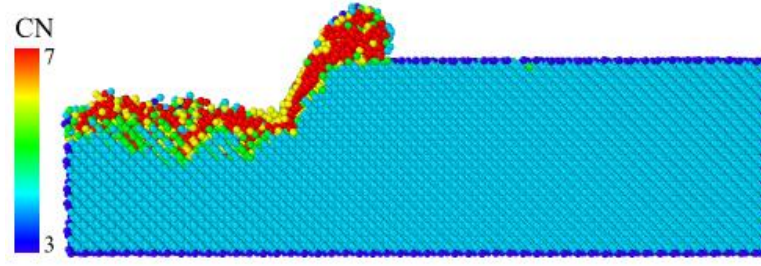


Fig. 6. Variation in the coordination number (CN) during nanomachining distance of 10 nm

The coordination number (CN) ranging from 3 to 7 were observed. A coordination number of 3 represents the dangling bonds on the surface of GaAs, 4 represents the zinc blende structure (GaAs-I), intermediate structures between CN 4, CN 6 and CN7 highlights high pressure phases including possible rocksalt structure (GaAs-II), and amorphous structures respectively. It can thus be considered that the occurrence of the high-pressure phase transformation (HPPT) governed the ductility of GaAs during nanometric cutting. More interestingly, a comparison of hydrostatic stress distribution and CN distribution revealed that a significant amount of rocksalt structures (GaAs-II, shown as yellow color atoms in Fig. 6) were generated at the peak hydrostatic stress area, i.e., primary shear slip zone and the rake face of the diamond tip. Correspondingly, this result signified the evolution of phase transformation of GaAs was in accordance with that of high-pressure during the nanomachining process.

3.3 Stress state of the diamond tip

It has been reported by some researchers that the cubic diamond lattice structure can transform to hcp graphite lattice structure under high shear stress state [57][58]. Therefore, the high shear stress is a considerable factor to investigate the underlying

mechanism for the graphitization of the diamond tip. The evolution of stresses on the diamond tip estimated from the simulations is shown in figure 7.

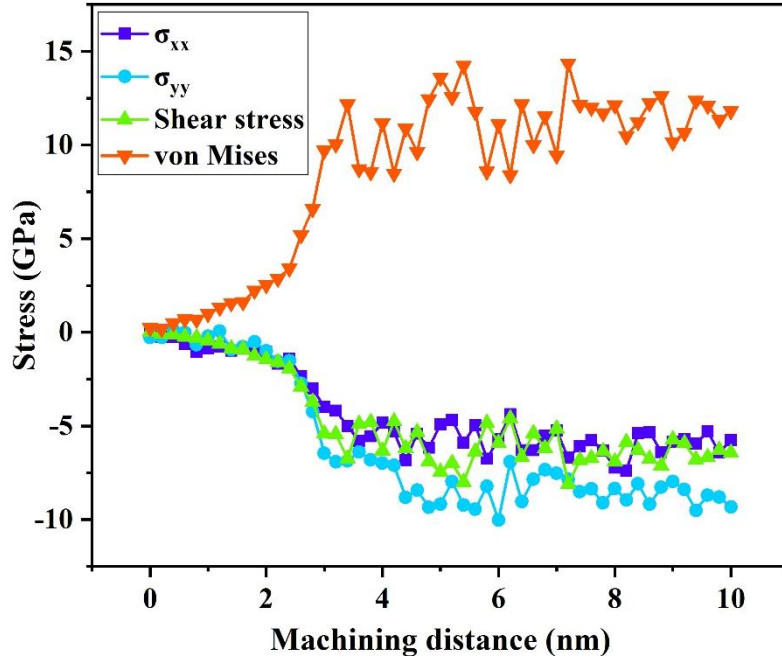


Fig. 7. The stresses variation on the diamond tip

The average shear stress was computed against machining distance, as illustrated in Fig. 7. It was found that the average shear stress fluctuated at a magnitude of 7 GPa. In addition, as shown in Fig. 8, after the cutting distance of 10 nm, the peak shear stress was mainly located at the apex of the diamond tip. The magnitude of stress was much lower than the shear stress of 95 GPa that was required to induce the cubic diamond lattice instability to the hcp graphite lattice structure [59]. This indicates that besides shear some other factor also contributes to causing the graphitization of diamond tip during nanometric cutting of single crystal GaAs.

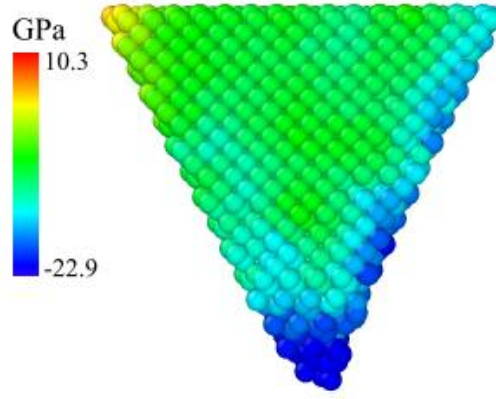


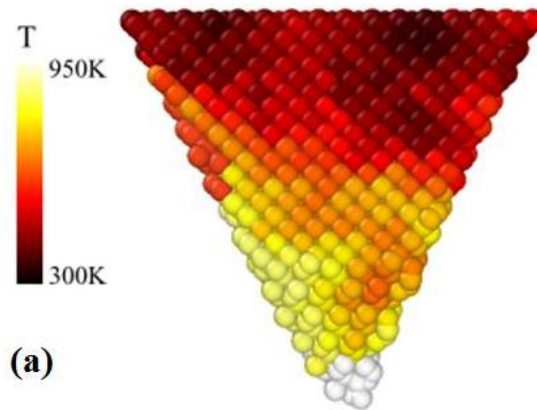
Fig. 8. The distribution of shear stress at the diamond tip after the cutting distance of 10 nm

As earlier shown in figure 7, the stress components σ_{xx} and σ_{yy} variation kept fluctuating between -6 GPa and -10 GPa indicating the compression state of the diamond tip. The von Mises stress was computed over the simulation period, which is employed to predict the yielding and shape distortion of the GaAs material [60][61]. The average von Mises stress of the diamond tip experienced a relatively stable stage at around 12.5 GPa between the cutting distances 4 nm and 10 nm.

3.4 Thermal state of the diamond tip

Another significant potential factor that may cause graphitization of the diamond tip is the cutting heat. The average temperature of the diamond tip dramatically increased at the start of the nanometric cutting, as illustrated in Fig. 9, and then kept steady at approximately 600 K. The stable average temperature rise of the diamond tip with the increase of cutting distance might be attributed to the quick heat dissipation thanks to the high thermal conductivity of diamond [62]. Another observation from Fig. 9(a) was that the locally highest temperature around 950 K was concentrated at the apex of the diamond tip, which was also the place of initiation of graphitization. It is known that

the minimum temperature required to cause the graphitization of diamond is about 940 K under anaerobic conditions [63], which is basically the same as the high temperature of the apex of the diamond tip based on the simulation results with the same simulation environment. Therefore, it implies that the high temperature at the apex of the diamond tip might weaken the cohesion energy of the C-C bond and further lead to the reorganization of the C-C bond with the advancement in the nanometric cutting [55][64]. Correspondingly, cutting heat appears to have triggered the sp^3 - sp^2 hybridization transition of the diamond tip.



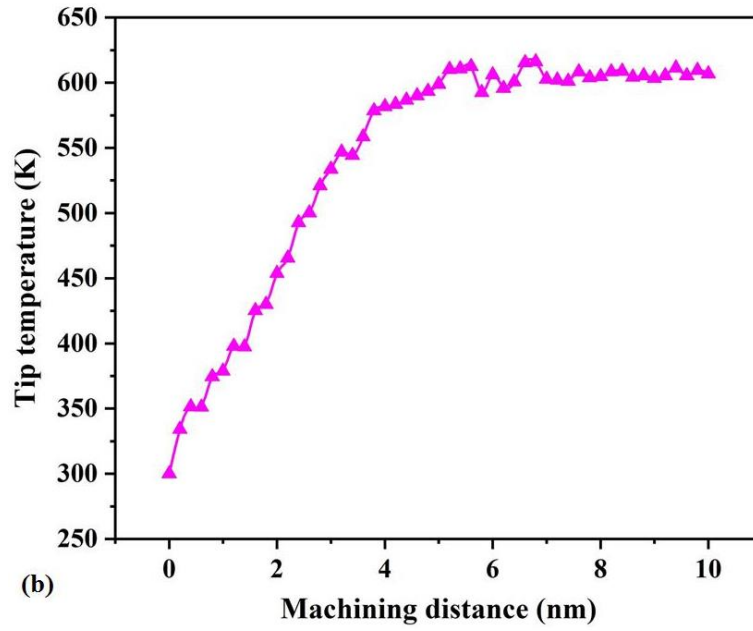


Fig. 9. (a)The temperature distribution of diamond tip at nanomachining distance of 10 nm. (b)The average temperature evolution of diamond tip

3.5 Graphitization conversion rate of diamond tip

It is well known that diamond has a coordination number of 4 while graphite has a coordination number of 3. In this paper, the graphitization conversion rate of the diamond tip was quantified by the division of the number of graphitized atoms to the total atom number (13107 atoms) of the diamond tip. To eliminate the influence of dangling bonds (292 atoms) of the diamond tip surface, the total number of graphitization atoms was calculated via the recorded number of CN 3 atoms deducting the initial number of dangling bonds atoms. This is akin to the research methodology that has previously been followed by one of the co-authors of this paper [29].

The graphitization conversion rate of the diamond tip is illustrated in Fig. 10. The trend of the graphitization conversion rate of the diamond tip could be split into two regions

during the course of nanomachining. The initially growing graphitization region implied that the initial rapid wear up to a cutting distance of 14 nm was more severe. Later stages of wear were somewhat steady. A similar trend of wear of diamond tool was found during the diamond turning of silicon [65] and silicon carbide [29] although the final rapid tool failure stage and final rapid wear stage did not reach in the MD simulation due to the short cutting distance. Therefore, this curve suggested that the continuous graphitization of diamond tip might lead to tool failure during nanometric cutting of GaAs.

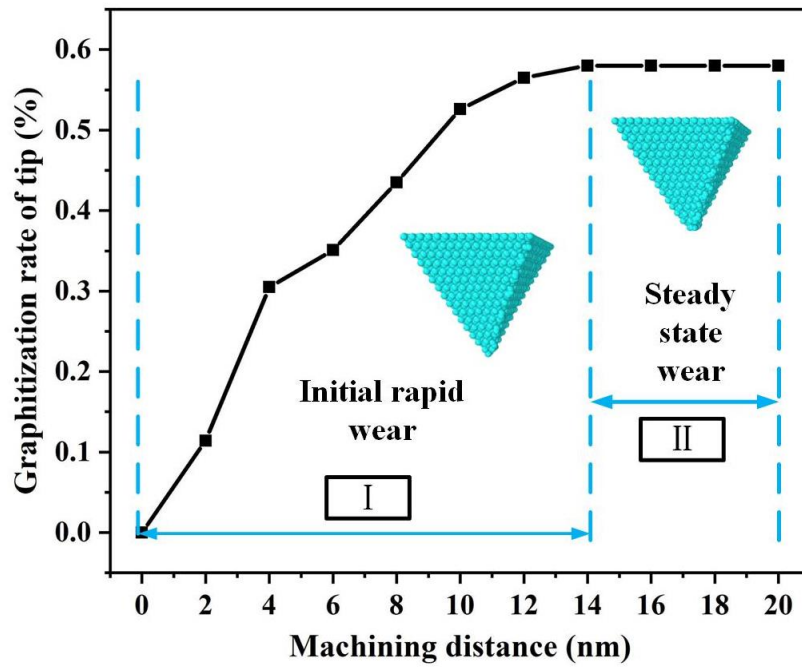


Fig. 10. The graphitization conversion rate of diamond tip

4. Conclusions

In this work, the wear mechanism of the diamond tip against GaAs during the AFM tip-based nanomachining process was investigated by using MD simulations. Based on the

aforementioned discussions, the main conclusions could be as follows:

1. sp^3 - sp^2 hybridization transition leading to graphitization of diamond is the dominant wear mechanism of the diamond tip during nanometric cutting of GaAs.
2. The root cause of diamond tip graphitization during nanometric cutting of GaAs was found to be the cutting heat rather than the high shear at the contact interface in the cutting zone.
3. Some indications of high-pressure phase transformation of the GaAs were obtained by using the coordination number method within the simulation and this aspect suggested a high likelihood of the phase transition of GaAs from zinc blende structure (GaAs-I) to rocksalt structure (GaAs-II) during the nanomachining process.
4. The variation in the variation of the graphitization conversion rate of the diamond tip against the cutting distance showed two distinct phases of wear with the initial stage suggesting severe wear followed by steady-state wear.

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Data Availability Statement

All data underpinning this publication are openly available from the University of Strathclyde KnowledgeBase at <https://doi.org/10.15129/01270576-2ee6-452c-927a-b5a0a89a0921>.

Conflict of Interest: The authors declare that they have no conflict of interest.

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