***Biophysical investigation into the antibacterial action of modelin-5-NH2***

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**Abstract**

Modelin-5-CONH2 (M5-NH2)is a synthetic antimicrobial peptide, which was found to show potent activity against *Bacillus subtilis* (Minimum lethal concentration = 8.47 µM) and to bind strongly to membranes of the organism (Kd = 10.44 µM). The peptide adopted high levels of amphiphilic α-helical structure in the presence of these membranes (> 50 %), which led to high levels of insertion (Δπ ≥ 8.0 mN m-1). M5-NH2 showedhigh affinity for anionic lipid (Kd = 7.46 µM) and zwitterionic lipid (Kd = 14.7 µM), which drove insertion into membranes formed from these lipids (Δπ = 11.5 and 3.5 mN m-1, respectively). Neutron diffraction studies showed that M5-NH2 inserted into *B. subtilis* membranes with its N-terminal residue, L16, located 5.5 Å from the membrane centre, in the acyl chain region of these membranes, and promoted a reduction in membrane thickness of *circa* 1.8 Å or 5 % of membrane width. Insertion into *B. subtilis* membranes by the peptide also promoted other effects associated with membrane thinning, including increases in membrane surface area (Cs-1 decreases) and fluidity (ΔGmix > 0 to ΔGmix < 0). Membrane insertion and thinning by M5-NH2 induced high levels of lysis (> 55%), and it is speculated that the antibacterial action of the peptide may involve the toroidal pore, carpet or tilted-type mechanism of membrane permeabilization.

Introduction

The emergence of microbial pathogens with multiple drug resistance (MDR) has rendered bacterial infections as one of the leading global causes of mortality, precipitating a serious public health issue in countries across the world 1. Due to the gravity of the threat posed by MDR pathogens, the World Health Organization (WHO) has ranked MDR as a priority issue, which if not remedied, could usher a return to the pre-antibiotic era when many common diseases were untreatable 2. More recently, major reports by the WHO and the UK Government have called for harmonized, global action to counter the threat posed by MDR pathogens, which includes the identification of new antibiotics and the development of novel antimicrobial strategies 3, 4. Most recently, the WHO took the unprecedented step of publishing a list of the twelve MDR bacteria that pose the greatest threat to human health with the aim of catalysing the development of crucially needed, new antibiotics to kill these ‘priority pathogens’ 5. For example, *Acinetobacter baumannii* is listed and whilst in the early 1970s, the organism was largely unknown and recognized as an opportunistic, nosocomial pathogen, primarily infecting critically ill or immunocompromised patients it was susceptible to most antibiotics. By the 1990s, it was increasingly becoming established as an emergent, opportunistic pathogen with MDR 6 and today, this organism has acquired resistance to virtually all antibiotics capable of treating Gram-negative bacteria, including carbapenems, which are ‘last-line’ β-lactams used to treat infections caused by these organisms 7. More recently, strains of carbapenem-resistant *A. baumannii* have emerged with resistance to colistin, which, due to its neurotoxicity and nephrotoxicity, was considered to be the ‘last resort’ drug capable of treating infections caused by the organism 8. Currently, carbapenem-resistant *A. baumannii* is regarded as the most difficult nosocomial pathogen to treat, which is reflected by its ‘critical’ priority and position at the top of the ‘priority pathogens’ list 5.

The call by the WHO and UK Governments for global action to combat the threat of MDR pathogens proposes a number of interventions and a major example is the therapeutic development of antimicrobial peptides (AMPs) 3, 4. These peptides are endogenous antimicrobials that have been identified in living things across the eukaryotic kingdom and have potent activity against not only bacteria, but also viruses, fungi, parasites and even cancer cells 9-12. A variety of resistance mechanisms to AMPs have been described in bacteria, which generally involve the inherent properties of these organisms rather than the acquisition of dedicated resistance genes 13. This intrinsic protection from AMPs is generally moderate in level and relatively non-specific in nature, which is reflected in the low frequency of resistant bacteria and the evolutionary persistence and success of these peptides 14. The evolution of bacterial resistance to AMPs has been demonstrated experimentally 15, but based on the pharmacodynamics of these peptides, it has been predicted that they are much less likely to select for resistant mutants *in vivo* than conventional antibiotics 16. Indeed, recent studies have suggested that the likelihood of bacteria developing resistance to AMPs is greatly reduced when they are exposed to synergistic combinations of these peptides, as would be the case *in vivo* 17. These observations have led to the generally held view that bacterial resistance to AMPs is unlikely to approach the levels observed for conventional antibiotics 9, 10, although this view has been questioned 15. Other advantages of AMPs include high potency and selectivity, a broad range of targets, potentially low toxicity and low accumulation in tissues. These are highly desirable properties in the development of therapeutically useful antimicrobial agents 18. However, the full commercial exploitation of AMPs has been hindered for a variety of reasons, primarily their high cost of manufacture and their strong tendency to undergo proteolytic degradation *in vivo* 10, 12. In response, concerted attempts to develop AMPs by producing analogues, mimics, and *de novo*, synthetic forms of these peptides have been undertaken and currently, a number of these molecules are in clinical trials and therapeutic use 9, 10, 12.

A synthetic AMP with many of the properties needed for therapeutic development is modelin-5-CONH2 (M5-NH2), which is an α-helical peptide with anticancer activity 19-21 and broad range antimicrobial action 20, 22-25 that is accompanied by negligible lysis of erythrocytes from humans and other mammals 26. The peptide is effective against a variety of Gram-negative bacteria, including clinical isolates 22, and its ability to kill the archetypical member of these organisms, *Escherichia coli*, has been well characterised, appearing to involve a carpet-type mechanism 24, 25. M5-NH2 also possesses activity against Gram-positive bacteria 22, 23, *Bacillus subtilis*, which is generally taken as the model organism for this bacterial class and which forms the focus of the present study 27. The peptide was found to possess potent activity against this organism and a range of biophysical techniques, including neutron diffraction, lipid monolayers and CD spectroscopy, are used to investigate the antibacterial mechanism involved. This mechanism appears to involve the ability of anionic lipid in *B. subtilis* membranes to promote the adoption of high levels of amphiphilic α-helical structure by M5-NH2. This amphiphilic secondary structure then appears to drive the penetration, thinning and lysis of the organism’s membranes by the peptide. It is speculated that this lysis involves one of the following mechanisms of action: toroidal pore, carpet and tilted-type.

**Experimental**

**Materials**

M5-NH2 (KLAKKLAKLAKLAKAL-CONH2) and d-M5-NH2-L16 (KLAKKLAKLAKLAKAL(d16)-CONH2) were supplied by Pepceuticals (Leicestershire, UK), and were purified by HPLC to purity greater than 95%, confirmed by MALDI mass spectrometry. All buffers were prepared using ultra-pure water (resistivity 18 MΩ cm). The phospholipids 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoglycerol (POPG) and 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoethanolamine (POPE) were purchased from Avanti Polar Lipids (Alabaster, AL). Ringer’s solution (RS), Nutrient broth (NB) and Nutrient agar (NA) were purchased from Thermo Fisher Scientific (Leicestershire, UK). HPLC grade solvents were obtained from VWR International Ltd (Lutterworth, UK) and all other regents were purchased from Sigma-Aldrich Company Ltd. (Dorset, UK).

Methods

The theoretical analysis of M5-NH2. The sequence of M5-NH2 was modelled as a two-dimensional axial projection assuming an angular periodicity of 100° according to the helical wheel analysis of Schiffer and Edmundson 28. The amphiphilicity of this α-helical arrangement was quantified as < µH >, according to hydrophobic moment methodology using a moving window of eleven residues and the normalized consensus hydrophobicity scale of Eisenberg 29. The potential of M5-NH2 to form a tilted peptide was investigated by amphiphilic profiling using - < µH >, with a moving window of seven residues and the normalized consensus hydrophobicity scale of Eisenberg 30.

The antibacterial activity of M5-NH2. The antibacterial activity of M5-NH2 was determined using *B. subtilis* (strain NCIMB 1671), which was taken from a frozen stock (-80 °C), inoculated into 10 ml aliquots of sterile NB and incubated in an orbital shaker (100 rpm and 37˚C) until the exponential phase (OD = 0.6; λ = 600 nm) was reached. This culture was then centrifuged at 15000 × g and 4 °C for 10 min using a bench top centrifuge and the resultant cell pellet washed 3 times in 1 ml aliquots of 25% strength RS, before being resuspended in 1 ml of 25 % strength RS. For the standard assay, these samples were further diluted with 25 % strength RS to achieve a bacterial density of 106 CFU ml-1. Then, 10 μl aliquots of these cell suspensions were taken, inoculated with 1 ml of M5-NH2 in RS to give final peptide concentrations of 3.90 μM to 1 M and incubated overnight at 37 °C. As a control, cultures of *B. subtilis* were similarly treated but in the absence of M5-NH2. After incubation, 10 µl aliquots of the control samples and bacterial cultures that had been treated with M5-NH2 were surface spread onto an NA plate and incubated at 37 °C for 12 hours. After incubation, the plates were viewed and the lowest peptide concentration yielding no bacterial growth was identified as the minimal lethal concentration (MLC) of M5-NH2. These experiments were repeated 4 times and the average MLCs determined 25.

The conformational analysis of M5-NH2. The conformational preferences of M5-NH2 in the presence of small unilamellar vesicles (SUVs), formed from POPE, POPG, or lipid mixtures, were analysed using a J-815 spectropolarimeter (Jasco, UK) at 20 °C with the vesicles acting as model *B. subtilis* membranes. These SUVs acting as model *B. subtilis* membranes were formed from mixtures of POPG and POPE in the molar ratio 70:30 or total lipid extracts of the organism’s membranes, which were dissolved in chloroform. These various lipid solutions were then dried under N2 gas, vacuum-dried for 4 hours, and the resulting lipid films rehydrated using 1 × PBS (pH 7.5). Rehydrated samples were then vortexed for 5 min and the resulting lipid suspensions sonicated for 30 min using a Soniprep 150 (ISTCP, USA) sonicator until clear, followed by three cycles of freeze-thawing. The resulting solutions of SUVs were then extruded 11 times through a 0.1 μm polycarbonate filter using an Avanti (UK) polar lipids mini-extruder apparatus and diluted 10-fold using 1 × PBS (pH 7.5). These diluted SUVs were then mixed with stock M5-NH2 solution (final concentration of 0.1 mg ml-1) to give samples with a peptide:lipid molar ratio of 1:100. Both in the presence and absence of these SUVs, far-UV CD spectra were collected for M5-NH2, where four scans per sample were obtained using a 10 mm path-length cell. Each scan was performed over a wavelength range of 260 to 180 nm at 0.5 nm intervals employing a bandwidth of 1 nm and at a speed 50 nm min-1 31. For all spectra acquired, the baseline acquired in the absence of peptide was subtracted and the percentage α-helical content of M5-NH2 estimated using the CDSSTR method (protein reference set 3) from the DichroWeb server 32. These experiments were repeated 4 times and the percentage α-helicity was averaged.

The interaction of M5-NH2 with lipid monolayers. Insertion experiments were carried out at constant area to quantify the interaction of M5-NH2 with lipid monolayers using a 601M Langmuir trough (Biolin Scientific\KSV NIMA, Coventry, UK). Chloroformic lipid solutions (1 mg ml-1), which contained POPE, POPG, mixtures of POPG and POPE in the molar ratio 70:30 or lipid extracts from membranes of *B. subtilis*, were spread drop-wise onto an air-buffer (1 × PBS, pH 7.4) interface using a Hamilton syringe. The solvent was allowed to evaporate and, then, monolayers were compressed by 2 moveable Derlin barriers with a velocity of 50 mm min-1 to a starting surface pressure of 30 mN m-1, which corresponds to that generated by the packing density of naturally occurring cell membranes 33. Once the monolayer was stable at a starting pressure of 30 mN m-1, M5-NH2 was injected underneath the monolayer to give a final peptide concentration of 6 µM in the subphase (1 × PBS pH 7.4). The surface area of monolayers was kept constant via a built-in controlled feedback system, and surface pressure increases were monitored by the Wilhelmy method using a Whatman’s CH1 filter paper plate and microbalance 31. All experiments were carried out at 20 °C and repeated 4 times.

Thermodynamic analysis M5-NH2 with lipid monolayers. Compression isotherms were generated from monolayers formed from POPG and POPE in the molar ratio 70:30. Chloroformic solutions of these lipid molecules (2.5 × 1015) were spread onto a buffer subphase (1 × PBS, pH 7.4), the solvent allowed to evaporate for 10 min and the monolayer left to stabilize for a further 20 min. The trough barriers were then closed at a speed of 0.22 nm2 min-1 until monolayer collapse pressure was achieved. Surface pressure changes were monitored and plotted as a function of the area per lipid molecule. Corresponding experiments were then performed except that M5-NH2 was introduced into the subphase to give a final peptide concentration of 6.0 µM. All experiments were carried out at 20 °C and repeated 4 times 31.

Thermodynamic analysis of these isotherms was undertaken and compressibility moduli (Cs-1) determined to provide a measurement of the compressional elasticity of monolayers and hence information about the packing of their component molecules. Cs-1 was computed according to equation 1 [39]:

Cs-1 (1)

where π is surface pressure of the monolayer and A represents the area per lipid molecule in the monolayer.

Thermodynamic analysis of these isotherms was also used to determine the Gibbs free energy of mixing (ΔG*mix*) of monolayers, which provides a measure of the relative stability associated with the miscibility energetics of their pure lipid components. Thermodynamically stable and thermodynamically unstable monolayers are indicated by negative and positive values of ΔG*mix* respectively 33. ΔG*mix* was computed according to equation 2:

(2)

where A1,2, is the molecular area occupied by the mixed monolayer, A1, A2 are the area per lipid molecule in the pure monolayers of component 1 and 2, X1, X2 are the molar fractions of the components and π is the surface pressure. Numerical data were calculated from the compression isotherms according to the mathematical method of Simpson 34.

The membranolytic ability of M5-NH2. The membranolytic ability M5-NH2 was determined using a dye release assay 25. Chloroformic solutions of lipid (7.5 mg ml−1) in the form of either, POPG and POPE in the molar ratio 70:30 or total lipid extracts from membranes of *B. subtilis*. These solutions were dried under N2 gas and kept under vacuum for at least 12 hours to ensure complete removal of the solvent. The resulting lipid films were then hydrated with 1 ml of 5.0 M 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) (pH 7.5) containing calcein (70.0 mM) and the suspension vortexed for 5 min before being sonicated for 30 min, which was followed by 3 cycles of freeze-thawing to maximize calcein encapsulation. Vesicular encapsulated calcein was then separated from the free dye by elution with HEPES (5.0 mM, pH 7.5) down a Sephadex G75 column (SIGMA, UK), which had been rehydrated overnight in HEPES (20.0 mM, pH 7.5), NaCl (150 mM) and Ethylenediaminetetraacetic acid (EDTA, 1.0 mM). The calcein release assay was performed by combining 25 μL of vesicular encapsulated calcein with 50 μL of 10 µM M5-NH2, which was then made up to a final volume of 1 ml with 20.0 mM HEPES, 150.0 mM NaCl and 1.0 mM EDTA (pH 7.4). The fluorescence intensities of calcein was monitored at 20°C using an FP-6500 spectrofluorometer (JASCO, UK), with an excitation wavelength of 490 nm and emission wavelength of 520 nm. The fluorescence intensity induced by the addition of 10 μL of triton X-100 (10 %, v/v) to vesicular encapsulated calcein was taken to represent 100 % dye release and was used to calculate the relative % of calcein released from vesicles by M5-NH2. All experiments were repeated 4 times.

The membrane binding properties of M5-NH2. A fluorescent probe assay was used to evaluate the ability of M5-NH2 to bind to membranes 25. SUVs formed from POPG and POPE in the molar ratio 70:30 were prepared and fluorescein-phosphatidylethanolamine (FPE, 0.5 M) was added to the organic solvent before drying under vacuum overnight to create a lipid film. These films were then hydrated with Tris-HCl (10.0 mM, pH 7.4) and EDTA (1.0 mM), followed by freeze-thawing 5 times and extrusion 11 times through an Avanti mini-extruder apparatus containing a 0.1 µm polycarbonate filter. Fluorescence was recorded using an FP-6500 spectrofluorometer (JASCO, UK), with an excitation wavelength of 492 nm, an emission wavelength of 516 nm, and excitation and emission slits set to 5 nm. To investigate the binding of peptide to lipid vesicles, M5-NH2 in the range 0 to 325 µM, was added to the FPE-labelled SUVs and the fluorescence monitored. The change in fluorescence (ΔF) was then determined as the fluorescence of FPE-labelled vesicles in the presence of peptide minus that of FPE-labelled vesicles in the absence of peptide. These ΔF values were then plotted against the concentration of M5-NH2 and then fitted by non-linear least squares analysis to equation 3:

(3)

Where [A] is the concentration of M5-NH2, ΔF is the change in fluorescence, ΔFMax is the maximum change in fluorescence and Kd, is the binding coefficient of the peptide. All experiments were repeated 4 times and the average value of ΔF calculated.

Neutron diffraction localization of M5-NH2 in lipid bilayers. Neutron diffraction experiments to determine the disposition of M5-NH2 in lipid bilayers were carried out on the V1 neutron membrane diffractometer at BER II, Helmholtz-Zentrum für Materialien und Energie, Berlin 35. Chloroformic solutions (20 mg ml−1) containing POPG and POPE in the molar ratio, 70:30 were prepared. Aliquots (1 ml) of these lipid solutions, and these solutions containing either protonated M5-NH2 (3 % molar), or this peptide deuterated at leucine 16 of its primary structure, d-M5-NH2-L16 (3 % molar), were individually deposited onto quartz microscope slides using an artist’s airbrush. The slides were then placed in a vacuum desiccator for 12 h to remove all traces of chloroform before rehydration for 24 h at 25 °C with relative humidity maintained at 98 % using Teflon water baths containing saturated potassium sulphate solution in 8 % 2H2O. The scanning procedure consisted of sequential θ scans around the predicted Bragg angle for each order. The rocking scans covered the Bragg position θ for the angle θ ± 2°. Diffraction patterns of the prepared samples were measured with up to five orders detected for each sample. The raw data from the two dimensional detector were summed to intensity versus 2θ using the V1 instrumental software 36. The commercial software IGOR Pro (version 4) was used for all further data analysis. The lamellar, or bilayer, spacing ‘d’ of each sample was calculated by least-square fitting of the observed 2θ values to equation 4, the Bragg equation:

Nλ = 2d sinθ (4)

where N is the diffraction order and λ is the neutron wavelength (4.52 Å). The integrated intensities were calculated based upon the Gaussian fit of the experimental Bragg reflections. Absorption correction and Lorentz factor were applied and their intensities square-rooted to produce structure-factor amplitudes. The phase assignment of each order and the relative scaling of the different data sets were determined by contrast variation in the aqueous atmosphere by adjusting the molar 2H2O/H2O ratio to 8 %, 20 %, and 50% 2H2O 36. The scattering length density profiles ρ(z) were then calculated for each sample using equation 5:

ρ(z) = ρ0 + 2/d ∑ ƒ(h) cos (2πhz/d) (5)

where ρ0 (z) is the integral density per unit length of the bilayer, ƒ(h) are the scaled structure factors and the second term describes the distribution in the scattering lengths across the bilayer.

Results

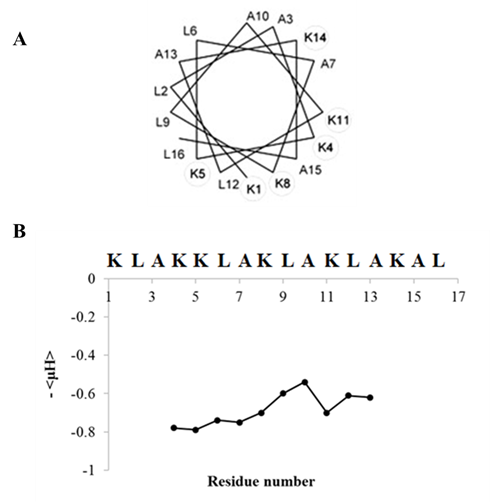
The theoretical analysis of M5-NH2. The sequence of M5-NH2 was modelled as a two-dimensional axial projection and the peptide displayed amphiphilicity with six lysine residues and the amide moieties on lysine 1 and leucine 16 generating positive charge on the hydrophilic face. (Fig. 1A) and suggesting a preference for binding to anionic membrane components 37. The amphiphilicity was quantified by the hydrophobic moment, < µH >, with a value of 0.75, which is typical of peptides active at the membrane interface 29 and consistent with an ability to partition into the membrane 37.

Fig. 1. Theoretical analysis of M5-NH2. M5-NH2 displayed a cationic α-helix, with a hydrophilic face rich in lysine residues and a hydrophobic face comprising alanine and leucine residues (Fig. 1A). The hydrophobic moment for this α-helix was *< µH >* = 0.75, indicating strong amphiphilicity and the potential for high levels of activity at a membrane interface 29. M5-NH2 showed an amphiphilic profile with - *< µH >* generally increasing over residues 4 to 13 in the N → C direction, indicating the potential for hydrophobicity gradient and tilted peptide formation. Values of - *< µH >* increased from *circa* – 0.8 to – 0.6, consistent with formation of a tilted peptide that occupies locations in the upper regions of the membrane (Fig. 1B) 30.

The potential of M5-NH2 to form a tilted peptide was investigated by amphiphilic profiling using -*< µH >*, which provides a graphical representation of the hydrophobicity gradients possessed by these peptides along their α-helical long axis. This asymmetric distribution of hydrophobicity is predicted to cause the parent peptide to penetrate membranes at a shallow angle of between 20° and 80°, thereby promoting a range of membrane destabilizing effects including the disturbance of lipid organisation and the compromise of bilayer integrity 30, 37. M5-NH2 exhibited a hydrophobicity gradient with values of - *< µH >* that generally increased from *circa* – 0.8 to – 0.6 in the N → C direction and which extended over residues 4 to 13 (Fig. 1B). Values of - *< µH >* of this order are consistent with tilted penetration by the peptide into the surface regions of the membrane 30, 37. This contrasts to the tilted peptide, maximin H5, whose hydrophobicity gradient is characterised by much lower values of - *< µH >* than those of M5-NH2 and appears to promote deep penetration of the central core region of membranes 31, 38.

The antibacterial and membranolytic activity of M5-NH2. The antibacterial activity of M5-NH2 was evaluated against *B. subtilis* using a standard assay, and the peptide exhibited potent action against this organism with an MLC of 8.47 µM. This level of activity is *circa* 10 to 15 fold stronger than that generally shown by the peptide towards other bacteria, indicating that B. subtilis is highly susceptible to the action of M5-NH2 19, 20, 22-25. For all AMPs, antibacterial activity involves membrane interaction 9, 10, 12 and the calcein release assay is frequently used to probe the structure / function relationships involved in these interactions 39. Use of this assay showed that M5-NH2 induced 59.3 % leakage of the dye in vesicles formed from POPG and POPE in the molar ratio 70:30 and 56.77 % leakage in those formed from lipid extracts of membranes from *B. subtilis*. These levels of calcein release show that M5-NH2 is able to induce high levels of permeability in these membrane mimics and are typical of strongly membranolytic AMPs, as reported for the antibacterial and anticancer action of maximin H5 31, 38. Taken with the high level of toxicity shown by M5-NH2 towards *B. subtilis*, these calcein release data also illustrate the general tendency of AMPs to reflect potency in biological action as efficacy in membranolysis 9, 10, 12.

The conformational analysis of M5-NH2. CD spectroscopy is commonly used to study conformational changes in AMPs, a primary determinant in their antimicrobial action 40. Use of this technique showed that for all SUVs studied, M5-NH2 displayed spectra with minima at 221 – 222 nm and 209 – 210 nm, and maxima at about 195 nm (Figs 3A and 3B), characteristic of α-helical structure 31, 38. Analysis of these spectra showed that M5-NH2 possessed 32.67 % α-helicity in the case of SUVs formed from POPE and 60.40 % α-helicity in the case of those formed from POPG (Figs 2A). In relation to lipid mimics of *B. subtilis* membranes, the peptide exhibited 53.40 % α-helicity in the presence of SUVs formed from POPG and POPE in the molar ratio 70:30 and 52.62 % α-helicity in the presence of those formed from lipid extract of the latter organism’s membranes (Fig. 2B).

M5-NH2 is known to be unstructured in aqueous solution 23 and in combination, these data indicate that the peptide generally undergoes a conformational rearrangement at a membrane interface to adopt α-helical structure. This is enhanced by the presence of PG species or anionic lipid in general and is consistent with previous findings 23, 25. Such interfacial conformational behaviour is typical of α-helical AMPs and facilitates the generation of amphiphilic structures that constitute the membrane interactive forms of these peptides 9, 10, 12.

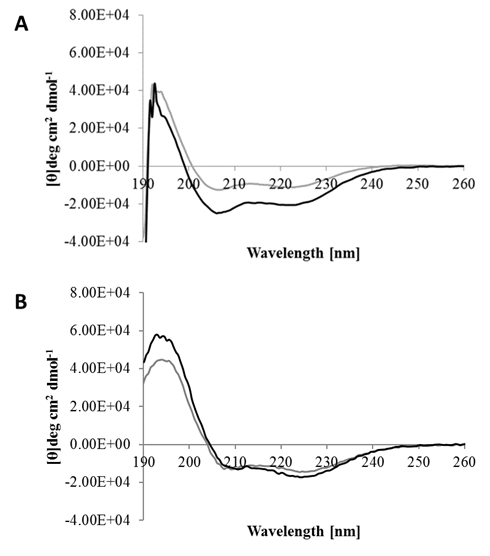
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Fig. 2. CD spectral analysis of M5-NH2. In the presence of all SUVs examined, M5-NH2 displayed spectra with minima at 221 – 222 nm and 209 – 210 nm and maxima at about 195 nm (Figs. 2A and2B), which is characteristic of α-helical structure 31, 38. In the case of SUVs formed from POPE (grey) and POPG (black), the peptide possessed 32.67 % and 60.40 % α-helicity, respectively (Fig. 2A). In the presence of SUVs, which acted as lipid mimics of *B. subtilis* membranes, M5-NH2 possessed 52.62 % α-helicity in the case of those formed from lipid extract of the organism’s membranes and 53.40 % α-helicity in the case of those formed from POPG and POPE in the molar ratio 70:30 (Fig. 2B). These data indicated that the peptide has a general ability to adopt α-helical structure in the presence of membranes that is enhanced by PG species or the anionic lipid of bacterial membranes, a common feature in the action of many AMPs 37.

The interaction of M5-NH2 with lipid monolayers. Lipid monolayers are highly effective for investigating the interaction of AMPs with membranes when precise control over membrane composition and compression is required 33. Use of this methodology showed that M5-NH2 partitioned into lipid monolayers formed from either POPE or POPG with generally similar kinetics, exhibiting very rapid initial rates of insertion over *circa* five seconds (Fig. 3A). The peptide then inserted into these monolayers at a slower rate until maximal surface pressure changes were induced after approximately 500 seconds, indicating saturation with M5-NH2. However, these maximal surface pressure changes varied widely, with 3.5 mN m-1 observed in the case of monolayers formed from POPE, indicating relatively low levels of interaction by the peptide. In contrast, maximal surface pressure changes of 11.5 mN m-1 were observed in the case of monolayers formed from POPG, indicating high levels of interaction by M5-NH2 (Fig. 3A). In combination, these monolayer results clearly reflect the structural amphiphilicity of M5-NH2 (Fig. 1A) and reinforce the observation that the peptide has a strong interaction with either PG species or anionic lipid, in general

In relation to lipid mimics of *B. subtilis* membranes, M5-NH2 exhibited very rapid initial rates of insertion over *circa* five seconds in the case of both, monolayers formed from POPG and POPE in the molar ratio 70:30 and those formed from lipid extracted from the organism’s membranes (Fig. 3B). However, afterwards, the peptide took *circa* ten times longer to achieve maximal surface pressure changes in monolayers composed of lipid extract as compared to the former monolayers (Fig. 3B), which may relate to the greater compositional complexity of monolayers formed from naturally occurring *B. subtilis* membranes 41, 42. The maximal surface pressure changes achieved by M5-NH2 were 9.0 mN m-1 in monolayers formed from POPG and POPE in the molar ratio 70:30 and 8.0 mN m-1 in those formed from lipid extract of the organism’s membranes (Fig. 3B). These surface pressure changes are consistent with the apolar face of the α-helical peptide interacting with the monolayer acyl chain region. Concomitantly, association of the polar face of α-helical M5-NH2 with the monolayer head group region would be predicted to stabilise these hydrophobic interactions 33. Comparable results have been reported for other α-helical AMPs such as aurein 2.5, which kills bacteria using highly membranolytic mechanisms of action 43.

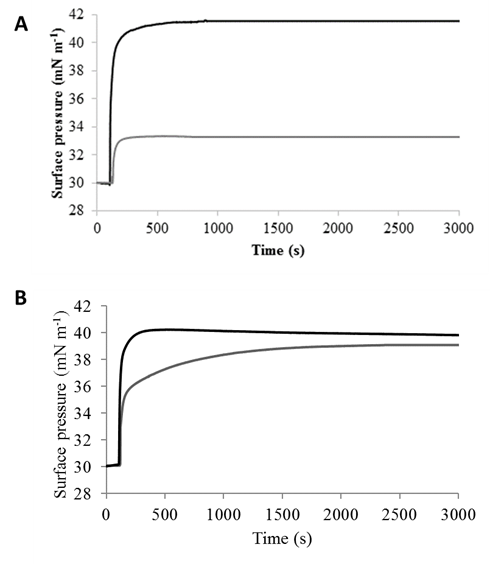


Fig. 3. The interaction of M5-NH2 with lipid monolayers. In the case of all monolayers examined, M5-NH2 showed high initial rates of insertion prior to achieving maximal surface pressure changes. These surface pressure changes were 3.5 mN m-1 in the case of monolayers formed from POPE (grey) and 11.5 mN m-1 in the case of those formed from POPG (black) (Fig. 3A), which clearly indicated an ability to interact with both zwitterionic and anionic lipid, reflecting the structural amphiphilicity of the peptide(Fig. 1A). For monolayers that acted as mimics of *B. subtilis* membranes, M5-NH2 displayed maximal surface pressure changes of 8.0 mN m-1 in the case of those formed from lipid extract of the organism’s membranes and 9.0 mN m-1 in the case of those formed from POPG and POPE in the molar ratio 70:30 (Fig. 3B). These data clearly indicated that the peptide has a general ability to partition into membranes that is enhanced by PG species or the anionic lipid of bacterial membranes, a common characteristic in the activity of many AMPs 37.

The membrane binding properties of M5-NH2. Another common approach to investigating the interaction of AMPs with membranes is to measure their lipid affinity using the fluorescence probe, FPE 44. This was used to evaluate the ability of M5-NH2 to bind to membranes (Fig. 4). In all cases examined, the binding of the peptide to SUVs followed hyperbolic kinetics, with fluorescence rapidly increasing as the concentration of M5-NH2 rose. At a concentration of *circa* 150 μM, changes in fluorescence became independent of peptide concentration, indicating that SUVs were saturated with bound M5-NH2 and no further binding of peptide could take place (Fig. 4). Analysis of these data indicated that the peptide bound to SUVs formed from POPG and POPE in the molar ratio 70:30 with a Kd value of 10.44 µM. This shows that M5-NH2 would have a high affinity for *B. subtilis* membranes, which is around tenfold stronger than that shown by the peptide for membrane mimics of *S. aureus* 23.

M5-NH2 also exhibited an affinity for both POPG and POPE (Fig. 4), which reflects the amphiphilic nature of the peptide (Fig. 1A). However, M5-NH2 bound POPG with a Kd value of 7.46 µM, which is *circa* two-fold stronger than that shown in the case of POPE where a Kd value of 14.7 µM was observed. In combination, these results clearly indicate that the peptide has a high affinity for PG species or anionic lipid, in general. These results also indicate that binding to *B. subtilis* membranes is likely to involve both hydrophobic interactions and a major contribution from hydrophilic associations.

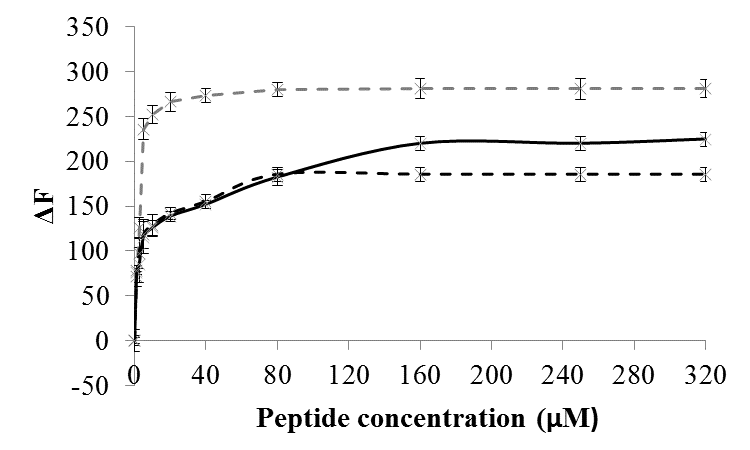
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Fig. 4. Lipid binding affinity analysis of M5-NH2. The lipid affinity of M5-NH2 was measured by its Kd value, which was derived from the maximum change in fluorescence induced by the peptide (ΔF) in FPE labelled SUVs. M5-NH2 bound SUVs formed from POPE (dashed black) with a Kd value of 14.7 µM and POPG (dashed grey) with a Kd value of 7.46 µM, demonstrating an affinity for both zwitterionic and anionic lipid, reflecting the structural amphiphilicity of the peptide (Fig. 1A). M5-NH2 bound SUVs formed from POPG and POPE in the molar ration 70:30 (black), which acted as lipid mimics of *B. subtilis* membranes, with a Kd value of 10.44 µM. These data indicated that the peptide has a general affinity for membranes that was enhanced by PG species or the anionic lipid of bacterial membranes.

Thermodynamic analysis of M5-NH2 interaction with lipid monolayers. Thermodynamic analyses of area-pressure isotherms derived from monolayers are frequently used to inform changes in the architecture and properties of membranes induced by their interaction with AMPs 33. Use of these analyses for monolayers formed from POPG and POPE in the molar ratio 70:30 (Fig. 5) showed that Cs-1 values increased with rising compression pressures, both in the presence and absence of M5-NH2 (Table 1) These Cs-1 values lay in the range 9.16 mN m-1 to 40.37 mN m-1 (Table 1), indicating that the corresponding monolayers were generally in the liquid expanded phase and were highly fluid across the compression pressure range of 5 mN m-1 to 20 mN m-1 33. For a given compression pressure, the presence of M5-NH2 led to reductions in the Cs-1 values of monolayers (Table 1), which indicated that their lateral pressure and lipid packing density had decreased. These changes are reflected as increased monolayer elasticity and fluidity and are consistent with the partitioning of the peptide into these *B. subtilis* membrane mimics 33. Similar results have been reported for other AMPs and, in these cases, changes in monolayer elasticity and fluidity have been primarily ascribed to membrane expansion *via* peptide – lipid head group interactions 45.

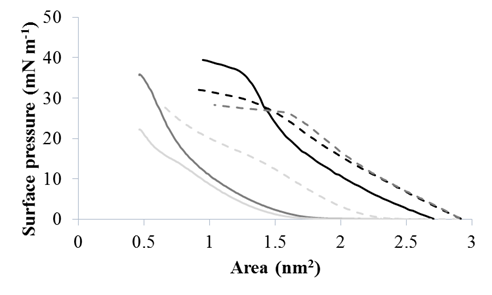
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Fig. 5 Thermodynamic analysis of M5-NH2 interaction with lipid monolayers. Compression isotherm analysis of monolayers formed from POPE (grey), POPG (light grey) and POPG and POPE in the molar ratio 70:30 (black) in the absence of M5-NH2 (solid lines) and presence of the peptide (dotted lines) were used to generate values of Cs-1 and ΔG*mix* for lipid mimics of *B. subtilis* membranes. Analysis of these data showed that, for a given compression pressure, the presence of M5-NH2 led to reductions in the Cs-1 values of these *B. subtilis* membrane mimics (Table 1), indicating that their lateral pressure and lipid packing density had decreased, consistent with partitioning of the peptide 33. These analyses also showed that for a given compression pressure, the presence of M5-NH2 changed ΔG*mix* values from > 0 to < 0 for *B. subtilis* membrane mimics (Table 1), indicating a thermodynamically stabilizing effect 33 that was consistent with the promotion of increased membrane fluidity *via* peptide – lipid head group interactions 45.

Compression isotherm analysis of monolayers formed from POPG and POPE in the molar ratio 70:30 (Fig. 5) was also used to determine values of Δ*Gmix* for these lipid systems (Table 1). In the absence of M5-NH2, these Δ*Gmix* values were > 0, increasing from 4.56 to 20.07 as the compression pressure rose from 5 mN m-1 to 20 mN m-1 (Table 1). These results indicated that there were energetically unstable interactions between the individual lipid components of these monolayers, consistent with a phase separation 33. However, in the presence of M5-NH2, these Δ*Gmix* values were < 0, decreasing from -2.99 to –13.84 across the same compression pressure range, indicating that there were attractive interactions between the individual monolayer components (Table 1). In combination, these data show that *B. subtilis* membrane mimics that were thermodynamically unstable (Δ*Gmix* > 0) were rendered thermodynamically stable (Δ*Gmix* < 0) by interaction with M5-NH2 33. Similar thermodynamically stabilizing effects are induced in these membrane mimics by other AMPs which were attributed to peptide – lipid head-group interactions that promoted increased membrane fluidity, paralleling our Cs-1 data (Table 1) 45. It is well established that increases in the elasticity and fluidity of membranes are often associated with thinning of the bilayer and are common characteristics of the antibacterial action of AMPs 46.

Table 1. Thermodynamic properties of monolayer mimics of *B. subtilis* membranes

|  |  |  |  |  |
| --- | --- | --- | --- | --- |
| Surface pressure  π (mN m-1) | POPG and POPE in the molar ratio 70:30  Cs-1 (mN m-1) | | POPG and POPE in the molar ratio 70:30  ∆G*mix* (kJ mol-1) | |
|  | - M5-NH2 | + M5-NH2 | - M5-NH2 | + M5-NH2 |
| 5 | 25.52 | 9.16 | 4.56 | -2.99 |
| 10 | 36.12 | 30.17 | 10.04 | -6.68 |
| 15 | 37.20 | 32.98 | 14.66 | -9.86 |
| 20 | 40.37 | 34.52 | 20.07 | -13.84 |

Neutron diffraction localization of M5-NH2 in lipid bilayers. Neutron scattering in conjunction with the substitution of hydrogen by its heavier isotope, deuterium, is frequently used to analyse the local structure, dynamics and interactions of multi-component, biomolecular systems 47. To assess the disposition of M5-NH2 in *B. subtilis* membrane mimics formed from POPG and POPE in the molar ratio 70:30, the N-terminal residue, L16, of the peptide was deuterated and its interaction with these membranes studied by neutron diffraction (Fig.  6), which is a form of elastic scattering 47. The lamellar d-spacings obtained for these lipid membranes were d = 53.6 ± 0.2 Å, in the absence of the peptide, d = 51.7 ± 0.2 Å in the presence of M5-NH2 and d = 52.6 ± 0.2 Å in the presence of d-M5-NH2-L16. These changes in lamellar d-spacings indicate that the thickness and hydration of the bilayers were affected by interaction with the peptide 48. The structure factors from Table 2 were employed to determine neutron diffraction density profiles for the interaction of M5-NH2 and d-M5-NH2-L16 with bilayers formed from POPG and POPE in the molar ratio 70:30 (Fig. 6). For these determinations, an 8 % 2H2O contrast was used to provide an H2O/2H2O ratio at which the mean neutron scattering length of the water mixture was zero 48. The experimentally determined scattering density profiles of samples containing either M5-NH2 or d-M5-NH2-L16, scaled to the sum of the respective structure factors, were subtracted to reveal the position of the deuterated L16 residue within the bilayer. A positive difference between these two profiles was indicated at a distance of 5.5 ± 0.1 Å from the membrane centre (Fig.  6). To determine the label position directly from the structure factors, a fit of a Gaussian model to the difference in structure factors in reciprocal space was performed 36.  The result indicated that the label position was at 5.7 ± 0.1 Å from the membrane centre, which is close to that obtained from the difference in experimentally determined density profiles. The difference in the fitted density profile, calculated with 5 structure factors, is less ‘wavy’ than the difference in the experimentally determined profiles, as in this case, the origin is solely the Fourier truncation error. These results locate L16 of M5-NH2 within the acyl chain region of the B subtilis membrane mimics.

For scattering density profiles, the water layer thickness of bilayers, dw, is defined by twice the distance of the unit cell edge to the position of half of the scattering length maximum, zw 36.  For example, the water boundary defined by zw for a leaflet of bilayers formed from POPG and POPE in the molar ratio 70:30 in the presence of d-M5-NH2-L16 is indicated in Fig. 6 by the thin vertical line. Analysis of Fig. 6 showed that for these bilayers, dw decreases from 9.2 ± 0.2 Å to 8.4 ± 0.2 Å in the presence of M5-NH2 and 8.6 ± 0.2 Å in the case of d-M5-NH2-L16. The distance between the density peaks corresponding to the lipid head-groups in scattering density profiles is indicative of the thickness of the bilayer 36 and in the case of those formed from POPG and POPE in the molar ratio 70:30, this distance was 36.8 ± 0.2 Å in the absence of the peptide (Fig.  6). This distance was decreased to 35.0 ± 0.2 Å in the presence M5-NH2 and to 35.06 ± 0.2  Å in the presence of d-M5-NH2-L16, signifying a reduction in the bilayer thickness of around 1.8 Å. Taken together, these data indicate that the peptide modulates structure in the acyl chain region of *B. subtilis* membranes and induces a membrane thinning effect; this is consistent with the thermodynamic data (Fig.  5). Comparable decreases in bilayer thickness have been reported for a number of other membrane thinning AMPs, such as magainin 2 and BP100 49.

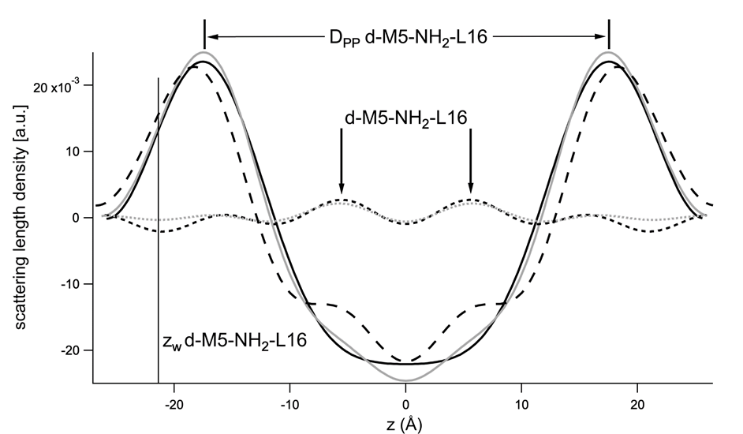


Fig.  6. Neutron diffraction studies on M5-NH2 / d-M5-NH2-L16 – membrane interactions. Neutron scattering density profiles were produced for a transection taken perpendicular to the surface of bilayers formed from POPG and POPE in the molar ratio 70:30, which acted as lipid mimics of *B. subtilis* membranes. These bilayers were either in the absence of peptide (dashed black), or in the presence of either M5-NH2 (black) or d-M5-NH2-L16 (grey), at 8 % 2H2O. The difference between the experimentally determined scattering density profiles of samples with either M5-NH2 or d-M5-NH2-L16 (dotted black), and a Gaussian model fitted to the difference in structure factors in reciprocal space (dotted grey) 36 revealed the position of the deuterated L16 residue within the bilayer. In relation to the membrane centre (z = 0 Å 48), a positive difference between the experimentally determined profiles was revealed at a distance of z = ± 5.5 ± 0.1 Å, and at a distance of z = ± 5.7 ± 0.1 Å for the Gaussian model fitted to the difference in structure factors in reciprocal space. In both cases, these differences are attributed to L16 of M5-NH2 residing within the acyl chain region of the bilayer. The water boundary, zw, for a leaflet of bilayers formed from POPG and POPE in the molar ratio 70:30 in the presence of d-M5-NH2-L16 is indicated by the thin vertical line. The position of the density peaks corresponding to the lipid head-group regions of these bilayers provides information regarding their thickness 36. The distance between these lipid head-groups, or density profile peaks (Dpp), was 36.8 ± 0.2 Å in the absence of peptide, which was decreased to 35.0 ± 0.2 Å in the presence M5-NH2 and 35.06 ± 0.2 Å in the presence of d-M5-NH2-L16, as is indicated above. In combination, these data indicate that the peptide modulates structure in the acyl chain region of *B. subtilis* membranes. This effect decreased the bilayer thickness by *circa* 1.8 Å, or 5 %, inducing a membrane thinning effect. Comparable decreases in bilayer thickness have been reported for a number of other membrane thinning AMPs 49.

Table 2. Experimental structure factors F(h), corrected and scaled, as described in the text

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| POPG and POPE in the molar ratio 70:30 | F(1) | F(2) | F(3) | F(4) | F(5) |
|  |  |  |  |  |
| + M5-NH2 | 0.498 ± 0.0017 | -0.254 ± 0.0040 | -0.215 ± 0.0030 | -0.033 ±0.0070 |  |
| + d-M5-NH2-L16 | 0.471 ± 0.0020 | -0.252 ± 0.0040 | -0.203 ± 0.0040 | -0.04 ± 0.0210 | 0.03 ±  0.0130 |

Discussion

By far the biggest group of naturally occurring AMPs are those that adopt amphiphilic α-helical conformations. This form of secondary structure has been used as a template for the production of numerous synthetic peptides with potential as antimicrobials 50. A number of these AMPs are based on a periodicity of lysine residues alternating with alanine and leucine residues that is designed to optimize the levels of amphiphilicity and α-helicity of the parent structure (LAK AMPs) 51. The use of these residues minimises production costs and LAK AMPs have been employed in a number of capacities, ranging from serving as the membrane disrupting domain of targeted pro-apoptotic peptides 52 to enhancing the antimicrobial action of sonosensitizers when conjugated with these molecules 53. LAK AMPs also serve as potent, broad range antibacterial agents 51 with activity against MDR pathogens such as *A. baumannii,* and the ability to synergize the action of conventional antibiotics against these pathogens 54. A major example of LAK AMPs is modelin-5-CONH2 (M5-NH2) and in the present study, we have sought to investigate mechanisms underpinning the activity of the peptide against *B. subtilis* 19-25. Here M5-NH2 was found to kill *B. subtilis* at micromolar levels of peptide (8.47 µM). Such levels of antibacterial activity are desirable when screening peptides for the potential to serve as leads in designing therapeutic AMPs 10.

It is well established that the antibacterial action of all known AMPs involves interaction with membranes and M5-MH2 showed a clear ability to interact with those formed from the native lipids of *B. subtilis* (Figs 3B, 4 and 5). However, to gain a deeper understanding of these interactions and to enable a more detailed data analysis in compression isotherm studies and neutron diffraction investigations (Figs 5 and 6), *B. subtilis* membranes were modelled by synthetic lipid mixes rather than bacterial lipid extracts. The majority of the native lipids in *B. subtilis* membranes are various phosphatidylglycerol (PG) species (70 %) with the remainder comprising a variety of phosphatidylethanolamine (PE) species 41, 42. Accordingly, synthetic lipid mimics of *B. subtilis* membranes were formed from the representative PG species, POPG, and the representative PE species, POPE, in the molar ratio 70:30. Control experiments established that M5-MH2 showed similar conformational behaviour and lipid interactivity between these synthetic lipid mimics and native lipid mimics of *B. subtilis* membranes (Figs 2 and 3). Whilst it is recognised that these synthetic systems lack a range of non-lipid components, they are well documented as providing information regarding the important role of lipid membrane components in relation to the mode of action used by AMPs 33.

In general, the interaction of AMPs with the bacterial membrane leads to the death of these organisms *via* membranolysis and a number of models have been presented to describe this process, such as the barrel stave, toroidal pore, carpet and tilted-type mechanisms 37. Based on similarities to these models, a synthesis of the data presented here permits a scheme to be constructed that describes the ability of M5-MH2 to kill *B. subtilis*. In the first step of this scheme, the strongly cationic peptide targets anionic components of *B. subtilis* membranes, including PG species. M5-MH2 then binds to these membranes with high affinity (Kd = 10.44 µM, Fig. 4) to adopt α-helical structure (> 50 %, Fig. 2B), the membrane interactive form of M5-MH2 for this organism. The peptide is strongly attracted to anionic lipid or PG species and the major driver of this conformational change is the preponderance of these latter species in the membranes of *B. subtilis*. These PG species show a much stronger propensity to induce α-helical structure in the peptide (> 60 %, Fig. 2A) as compared to PE species (< 40 %, Fig 2A).

In the next step of our scheme, the high levels of α-helical structure possessed by the peptide in the presence of the organism’s membranes (Fig. 2A) drive it to partition strongly into these membranes(Δπ ≥ 8.0 mN m-1, Fig. 3B) and to interact with both their head group and hydrophobic core regions (Figs 3B and 6). Neutron diffraction studies clearly showed that the N-terminal residue, L16, of M5-MH2 was located around 5.5 Å from the centre of *B. subtilis* membranes and in the acyl chain region of these membranes (Fig. 6). The underlying driver of the peptide’s insertion into *B. subtilis* membranes appears to be the high amphiphilicity of the α-helical structure adopted in the presence of these membranes (*< µH >* = 0.75, Fig. 1). Indeed, the strongly α-helical peptide showed very high levels of partitioning into membranes formed from PG species (Δπ = 11.5 mN m-1,Fig. 3A). However, the loss of amphiphilic α-helical structure by M5-MH2 (> 20 %, Fig. 2A) in the presence of membranes formed from PE species led to around a threefold reduction in the ability of the peptide to partition into these membranes (Δπ = 3.5 mN m-1,Fig. 3A). These early steps in the interactions of the peptide with *B. subtilis* membranes are typical of α-helical AMPs with activity at the interface and are common to most models used to describe the membrane interactions of thee peptide 37. It is generally accepted that the exothermic action of α-helix formation contributes strongly to the enthalpy driven mechanisms of membrane insertion used by these AMPs 55.

The partitioning of M5-NH2 into *B. subtilis* membranes promotes membrane thinning effects and this was directly demonstrated by neutron diffraction studies where insertion by the peptide was shown to decrease membrane thickness by *circa* 1.8 Å or 5 % of its width (Fig. 6). Insertion by M5-NH2 was also found to promote increases in the surface area (Cs-1 ↓, Table 1) and fluidity (ΔG*mix*> 0 → < 0, Table 1) of *B. subtilis* membranes, which is consistent with a thinning effect 46. Similar results have been reported for other α-helical AMPs, such as magainin 2, and in these cases, membrane thinning effects have resulted from the insertion of these peptides into the membrane with their α-helices orientated parallel to the lipid surface 49, 56. Several studies have suggested that M5-NH2 may also partition horizontally into bacterial membranes 23-25, which could help explain the membrane thinning effect observed for the interaction of the peptide with membranesof *B. subtilis* (Table 1, Fig. 6). Indeed, it would seem that horizontally partitioning into these membranes would be aided by the even distribution of positive charge along the α-helical long axis of the peptide (Fig. 1A) 23-25. Consistent with the high affinity of M5-NH2 for PG species (Kd = 7.46 µM, Fig. 4), adopting this orientation would allow the peptide’s lysine groups (Fig. 1A) to interdigitate with the head groups of these species (Fig. 6). Indeed, the membranes of this organism include very high levels of diphosphatidylglycerol 41, 42, and it is well established that anionic moieties in the head group of this PG species are far more accessible to AMPs than those of other bilayer lipids 57. Horizontally partitioning into the membranes of *B. subtilis*, would also help drive penetration of the peptide’s alanine and leucine residues (Fig. 1A) into the hydrophobic core region of the bilayer, given the affinity of M5-NH2 for zwitterionic lipid (Kd = 14.7 µM, Fig. 4). Neutron diffraction studies showed that the membrane thinning ability of the peptide involved modulating the structure of the acyl chain region by its N-terminal residue, L16 (Fig. 6). Presumably, this action would promote the relaxation of acyl chains and lower the order of these lipid tails, as reported for other membrane thinning AMPs 49, 56.

In the latter stages of our putative scheme for the ability of M5-MH2 to kill *B. subtilis*, membrane insertion and thinning effects by the peptide induce high levels of membranolysis (> 55 %), resulting in the death of the organism. Based on the data presented here, it is not possible to definitively assign the antibacterial action of M5-NH2 to any particular model of membrane interaction. Given the relatively short length of the peptide, it seems unlikely that it would be able to form a membrane spanning pore, such as the barrel stave model proposed for a number of AMPs 37. However, the mode of membrane partitioning shown here for the peptide is often associated with the toroidal pore and carpet models 37, and these models have previously been predicted to describe the antibacterial action of M5-MH2 23-25. According to the first of these models, horizontal insertion and thinning by the peptide would destabilize the integrity of *B. subtilis*, membranes and lead to lysis through the generation of transmembrane channels formed by the intercalation of reorientated M5-MH2 molecules with lipid.  In the second of these models, a high density of the horizontally orientated peptide would accumulate on the surface of *B. subtilis* membranes, which would promote thinning and dissolution of the bilayer *via* a detergent-like action that does not involve channel formation 37. Neutron diffraction studies have revealed other mechanisms for the lysis of bilayers by horizontally orientated AMPs involving membrane thinning but not leading to pore or channel formation, such as that recently described for aurein 1.2. The binding of the peptide to membranes induced membrane thinning, accompanied by a major redistribution of PG species in the outer leaflet that appeared to promote the formation of complexes between lipid and aurein 1.2 and the compromise of membrane integrity 58. A further possibility based on the architecture of M5-MH2 may be that the mode of membrane partitioning used by the peptide leads to the use of a tilted-type mechanism. The peptide shows the potential to form α-helical structure with a hydrophobicity gradient that increases towards the N-terminal residue, L16, of M5-MH2 (Fig. 1B) and this form of secondary structure is a defining characteristic of tilted peptides that promotes their oblique membrane insertion 30, 37. Clearly, the lysis of *B. subtilis* membranes by a tilted mechanism would be consistent with the residence of L16 in the acyl chain region of these membranes (Fig. 6). Lysine residues possessed by the peptide could also play a role in promoting tilted membrane penetration *via* the snorkelling mechanism, which would allow the amine groups of these residues to remain associated with the lipid head-group region of membranes. Concomitantly, their long hydrocarbon side chains extend into the membrane’s hydrophobic region, permitting deeper levels of penetration by M5-MH2, as reported for other AMPs that appear to use this structure to facilitate their antibacterial action 59. Taken in combination, these observations suggest that the use of a tilted mechanism by M5-NH2 to lyse the membranes of *B. subtilis* would involve roles for both PE species and PG species in relation to orientating the peptide and modulating the overall effect of its action on lipid organisation and membrane architecture. These observations also reinforce the view that to fully characterise the antimicrobial action of AMPs, it is necessary to consider the properties of both of these peptides and their target membranes 37.

As a final comment, it is interesting to note that *B. subtilis* shows historical similarity to *A. baumannii*, in that it was first isolated from soil over a century ago and is generally considered to be non-pathogenic 60. However, since the 1960s, strains of *B. subtilis* have been responsible for a number of serious infections, generally in individuals who are critically ill or immunocompromised 61, 62. A very recent study, reported that the organism is increasingly prevalent in hospitalized patients and is sufficiently equipped with virulence mechanisms to behave as a human pathogen. It has been recommended that the identification of *B. subtilis* and other members of this genus should be routinely performed, particularly when samples derive from patients who are immunocompromised or have a pre-existing disease condition 63. *B. subtilis* is also commonly found in probiotic formulations and there are now growing concerns about the potential of this organism and other species of its genus for virulence when used as a dietary supplement 62-64. Moreover, the resistance of *B. subtilis* to antibiotics has been demonstrated 62 and it is becoming increasingly clear that it possesses the ability to resist AMPs; this would seem to make the development of M5-MH2 as an agent against the organism timely 65, 66. Indeed, *B. subtilis* may only be one example of an emerging problem which, for example, appears to include bacteria of the genus, *Lactobacillus*, another group of Gram-positive organisms used as probiotics 67. These bacteria are generally considered to be harmless but strains are emerging with pathogenic relevance to elderly and immunocompromised patients 68, as in a recent case of acute cholecystitis caused by vancomycin-resistant *Lactobacilli* 69.

Conclusion

In summary, M5-NH2 is a synthetic antibacterial agent with very low levels of haemolysis 26, and here it has been shown that the peptide has a potent ability to kill *B. subtilis* *via* membranolytic modes of action. Potentially, this action could be facilitated by a number of mechanisms, particularly tilted peptide formation: indeed, the data presented here could support the use of a recently reported, novel, tilted-type antibacterial mechanism. Using this mechanism, AMPs accumulate in the outer leaflet of the bacterial membrane and form pore-like structures by inserting in an oblique orientation. The resulting shallow pores promote rapid membrane thinning, leading to progressive disintegration and exfoliation of the outer leaflet at the sites of pore formation, which then induces cell death 70.

Conflicts of interest

The authors declare no conflict of interest.

Author contribution

S.R.D, T.H, K.B. F.H and D.A.P. designed the research. K.B. synthesised the peptide. S.R.D and T.H performed research. S.R.D, T.H, and F.H analysed data. S.R.D, T.H, F.H and D.A.P. wrote the manuscript.

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